

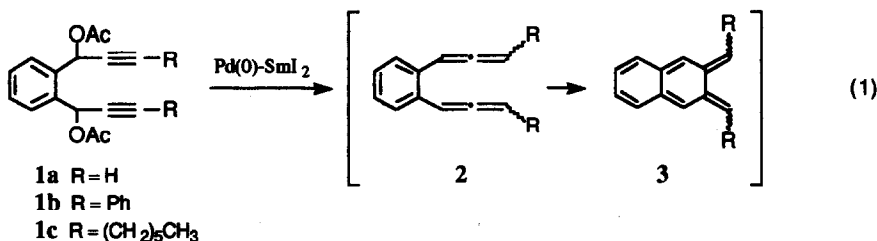
A NOVEL METHOD FOR THE GENERATION OF 2,3-NAPHTHOQUINODIMETHANES UTILIZING SAMARIUM(II) IODIDE-PROMOTED ALLENE SYNTHESIS

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Abstract: 2,3-Naphthoquinodimethanes are easily generated from *o*-bis(α -acetoxypropargyl)benzene derivatives with SmI₂ in the presence of palladium catalyst under mild conditions. The reactions of the quinodimethane intermediates were found to follow the different paths depending on the acetylenic substituents of substrates.

o-Quinodimethanes have been of intense interest not only from the structural and theoretical point of view but also because of their synthetic utility as effective dienes in Diels-Alder reaction for the construction of polycyclic compounds.¹ Therefore, a number of methods for the generation of these intermediates have so far been developed: e.g., (i) reductive, proton-induced, or anion-triggered 1,4-elimination reactions,² (ii) thermal or photochemical extrusion reactions and reverse Diels-Alder processes of cyclic precursors,³ (iii) ring-opening reactions of benzocyclobutenes and related compounds,⁴ etc. Among them, 6π -electron cyclization of *o*-diallenylbenzenes,⁵ for the formation of which strong bases were required, has attracted our attention because we recently developed a mild and convenient method for the formation of allenes from the corresponding propargylic acetates by using the Pd(0)-SmI₂-proton donor system.⁶ Thus, we attempted to apply our allene synthesis to the generation of *o*-naphthoquinodimethanes (3). (Eq. 1)



The precursors, bis-propargylic acetates (1a-c), were easily prepared from *o*-phthalaldehyde in two steps (LiC=CR; Ac₂O, DMAP) in 70-80% overall yields as mixtures of diastereoisomers (1:1, 1:2, and 1:9, respectively).

First, the reduction of *o*-bis(1'-acetoxy-2'-propynyl)benzene (1a) was examined and the postulated *o*-naphthoquinodimethane intermediate (3a)⁷ was tried to trap with various dienophiles. (Eq. 2) As expected, the

convenient one-pot reaction afforded the corresponding Diels-Alder adducts (4a-e)⁸ in satisfactory yields (Table 1) except for the reaction with dimethyl acetylenedicarboxylate (entry 5). A typical procedure is as follows: A SmI₂-THF solution (0.1 mol dm⁻³, 9.25 mL, 0.925 mmol) was added slowly to a solution of 1a (50.0 mg, 0.185 mmol), Pd(PPh₃)₄ (22.0 mg, 10 mol%), and 2,4-dimethyl-3-pentanol (58.0 μL, 0.407 mmol) in THF (4 mL) at -78°C. The mixture was stirred for 30 min at the same temperature and then gradually warmed to 0°C over several hours, while being monitored by TLC. After completion of the reduction, a solution of dimethyl fumarate (80.0 mg, 0.555 mmol) in THF (2 mL) was added and the whole mixture was stirred at 0°C for 2 h and then at room temperature for 12 h. To the reaction mixture was added hexane (2 mL) and silica gel (ca. 1 g). The resulting suspension was stirred for 10 min and then passed through a pad of silica gel. The filtrate was concentrated and purified by preparative TLC on silica gel to give 4b (33.2 mg, 60%) as a white solid.

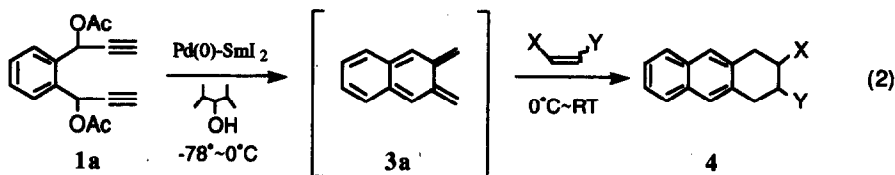


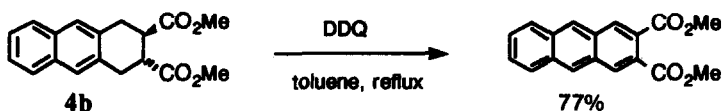
Table 1. One-Pot Conversion of 1a to the Cycloadducts (4a - 4e)^a

Run	Dienophile	Major product	Yield (%)
1			57
2			60
3			48 ^b
4			59
5			17

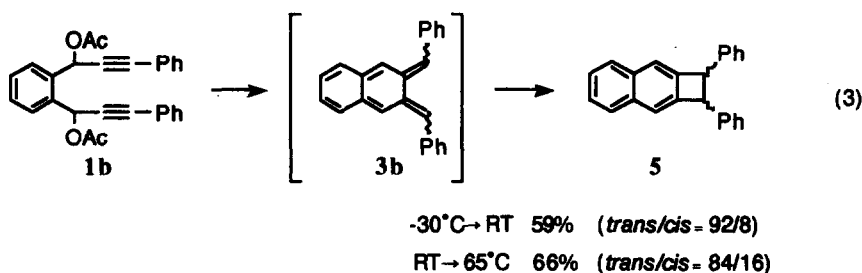
a) For the reaction conditions, see the text. b) A mixture of diastereomers: *cis/trans* = 72/28.

The concomitant formation of the cycloadduct (4b) in the reaction with dimethyl maleate (entry 3) would be due to the partial isomerization of the starting maleate to fumarate prior to the cycloaddition under the reaction conditions.⁹

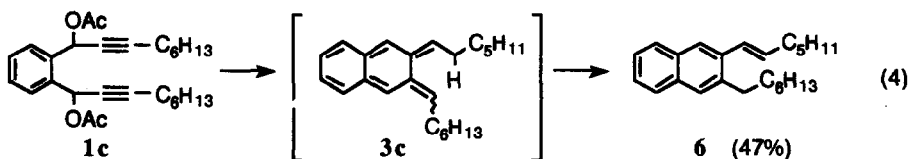
The cycloadduct **4b** could easily be converted into dimethyl 2,3-anthracenedicarboxylate⁸ in good yield.



The reduction of phenylacetylene derivative (**1b**) also proceeded smoothly, however, the corresponding [4+2] cycloadduct with methyl acrylate was not detected at all. A mixture of *trans*- and *cis*-diphenyl-naphtho[*b*]cyclobutene^{5c,8,10} (**5**, *trans/cis*=92/8) was isolated as the result of intramolecular [2+2] cycloaddition of the *o*-naphthoquinodimethane intermediate (**3b**).¹¹ (Eq. 3) When the reduction was performed at room temperature and then the mixture was stirred at 65°C for several hours without methyl acrylate, a mixture of the diastereoisomers (*trans/cis*=84/16) was obtained in 66% yield. The pure *trans*-**5** was, upon being allowed to stand for 30 days at room temperature, isomerized to give an equilibrium mixture of the stereoisomers (*trans/cis*=80/20),¹² indicating that cleavage and regeneration of the cyclobutene ring takes place thermally and/or photochemically probably through a radical process.^{5c-d,13}



Also, the reaction of bis-alkynic compound (**1c**) with methyl acrylate did not give the corresponding Diels-Alder adduct under the standard conditions; the major product was 1-alkenyl-2-alkylnaphthalene (**6**)⁸ which would be stemmed from the corresponding naphthoquinodimethane (**3c**) via [1,5] hydrogen shift.^{5d,14} (Eq. 4)

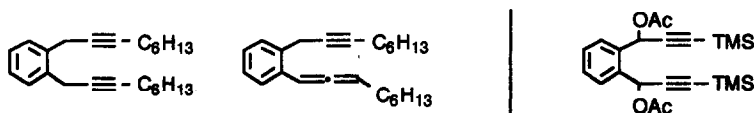


In conclusion, the present reductive method, which is characterized by ready availability of open chain precursors and mild reaction conditions, was found to provide a new route to *o*-naphthoquinodimethanes.¹⁵

References and Notes

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6. Tabuchi, T.; Inanaga, J.; Yamaguchi, M. *Tetrahedron Lett.* **1986**, *27*, 5237.
7. Sondheimer and his co-workers have tried to isolate the *o*-diallenylbenzene intermediate (2, R=H) and obtained it in >80% purity. See reference 5a.
8. Satisfactory spectral and analytical data were obtained.
9. In fact, dimethyl fumarate (ca. 15%) was detected in the recovered dienophile. It has been reported that the cycloaddition was at least 95% stereospecific when the reaction was conducted under neutral conditions; see reference 5a.
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11. It is not clear from the isomeric ratio of the products whether the process involves biradicaloid species because the precursor can be a mixture of meso and racemic diallennybenzenes. For mechanistic consideration, see references 5c and 5d.
12. MM2 calculations of *trans*-5 and *cis*-5 were carried out based on the reported data of the X-ray crystallographic analysis of naphtho[b]cyclobutene [Crawford, J. L.; Marsh, R. E. *Acta Cryst.* **1973**, *B29*, 1238]. The energy difference between the two isomers was calculated to be 0.99 kcal/mol. We thank Mr. Y. Yokoyama for carrying out the above calculations.
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14. In this reaction, a mixture of diacetylene and acetylene-allene derivatives (19:81) was also isolated as minor products (27% yield). Also, in the reduction of bistrimethylsilylacetylene derivative (1, R=TMS), the corresponding acetylenic compounds were selectively obtained and no cyclized product derived from the corresponding naphthoquinodimethane intermediate was detected.



15. A part of this work was presented at the 9th Symposium of Rare Earth Society of Japan, May, Kyoto (1991).

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